



Laboratory Exercises in Physical Chemistry

By Frederick Hutton Getman

Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1908 Excerpt: . run smoothly, the thermometer is inserted and the stirrer set in motion. As soon as the different parts of the calorimeter system have assumed a common temperature, which usually requires two or three minutes, the mercury will begin to rise at a uniform rate, and the readings of the initial or precombustion period may begin. The room temperature may have changed so much since the apparatus was last used that the thermometer must be reset. In that event the water should be stirred a little after the insertion of the bomb and its temperature determined with an ordinary thermometer, so that the actual temperature to which the zero of the reset thermometer corresponds may be known within half a degree. Temperature Changes in the System.-- If the calorimeter system were absolutely insulated thermally, only two temperature observations would be necessary for the determination of the...



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