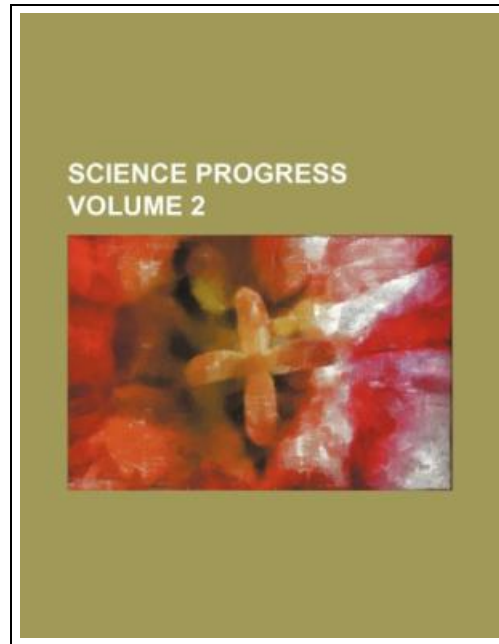


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1894 Excerpt: .an empirical formula, a concession to the weakness of the flesh. And certainly for the values of the constants assumed, p. 271, the formula gives results agreeing remarkably with those of Amagat. Does it not agree with them more closely than is necessary, considering that the calculated results are, in the matter of the constant c , only approximate? 30. The relation between kinetic energy and temperature. In the original or elastic sphere form of the theory we are compelled to treat kinetic energy of translation as identical with heat, and as a measure of temperature, because it is the only form of energy which we allow our system to possess. When we give our molecules many degrees of freedom, and as many forms of kinetic energy, the question arises, which of the corresponding forms of energy measures temperature? We quieted our consciences for a time by the law $m \cdot v^2$, saying that it did not matter $I n J$ which, because all that is required is a measure, and according to that law, either T or $x^2 + y^2 + z^2$, being always in the same ratio to one another, may be taken as the measure. But if we admit, as I think we must, that deviations from this law, small or great, must be always occurring, thanks to the ubiquitous ether or other disturbers of the peace, the question can no longer be evaded. On the whole, the tendency has been still to regard energy...



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